Facile Syntheses of Aporphine Derivatives

Martin H. Hedberg, Anette M. Johansson* and Uli Hacksell

Department of Organic Pharmaceutical Chemistry, Uppsala Biomedical Centre, Uppsala University, Box 574, S-751 23 Uppsala, Sweden

New and efficient synthetic routes, utilizing palladium-catalysed reactions, provide (R)-11-hydroxy-10-methylaporphine **2** and (R)-11-hydroxyaporphine **3** from natural morphine **4**.

(R)-Apomorphine 1 is a prototype dopamine (DA)-receptor agonist.¹ Recently, two derivatives of 1 have been shown to exhibit other interesting pharmacological profiles; (R)-11hydroxy-10-methylaporphine 2 appears to be a 5-hydroxytryptamine (5-HT) receptor agonist with selectivity for 5-HT_{1A}receptors² and (R)-11-hydroxyaporphine **3** has been characterized as a DA D₁-receptor antagonist.³ These aporphine derivatives were synthesized from natural morphine 4 in seven and five steps, respectively, with low to moderate overall yields.²⁻⁴ In this paper we report on two new, flexible and more efficient approaches producing 2 and 3 from 4 by use of palladium-catalysed synthetic methods (Schemes 1 and 2). The key-intermediate in both sequences is monotrifluoromethanesulfonate 5 {m.p. 123–124 °C, $[\alpha]_D^{21}$ –77.8 (c 1.0, MeOH)},† which is readily obtained from 4 by treatment with triethylamine and N-phenyltrifluoromethanesulfonimide in dichloromethane (CH₂Cl₂).⁵ Treatment of 4 with the more reactive trifluoromethanesulfonic anhydride gave a very low yield of 5.

In the first method (Scheme 1), monotrifluoromethanesulfonate **5** was converted to aporphine trifluoromethanesulfonate **6** {hydrochloride, m.p. 207–209 °C, $[\alpha]D^{21} - 48.0$ (*c* 1.0, MeOH)} by an acid catalysed rearrangement^{2.4.6} in concentrated methanesulfonic acid (MeSO₃H) followed by *O*-methylation of the C(11)-hydroxy group with diazomethane (CH₂N₂). Cross coupling of **6** with tetramethyltin [Me₄Sn] in the presence of tetrakis(triphenylphosphine)palladium [Pd(PPh₃)₄]‡ and lithium chloride (LiCl) in dimethylform-



† All compounds gave satisfactory combustion analyses and IR, ¹H NMR (270 MHz) and ¹³C NMR (67.5 MHz) spectroscopic data. Selected data: 3·HCl: ¹H NMR δ (CD₃OD) 2.88 (t, 1H), 3.10 (m, 1H), 3.16 (s, 3H), 3.45 (m, 3H), 3.77 (m, 1H), 4.33 (m, 1H), 6.86 (d, 2H), 7.10 (t, 1H), 7.15 (d, 1H), 7.35 (t, 1H), 8.40 (d, 1H). 5: ¹H NMR δ (CDCl₃) 1.90 (m, 1H), 2.11 (m, 1H), 2.28-2.40 (m, 2H), 2.44 (s, 3H), 2.62 (m, 1H), 2.70 (m, 1H), 3.09 (d, 1H), 3.39 (m, 1H), 4.21 (m, 1H), 5.02 (dd, 1H), 5.28 (m, 1H), 5.70 (m, 1H), 6.64 (d, 1H), 6.89 (d, 1H); ¹⁹F NMR δ (84.5 MHz; CDCl₃) -73.9 (s, 3F); IR v/cm⁻¹ (KBr) 3550 (OH). 6·HCl: ¹H NMR δ (CD₃OD) 2.93 (t, 1H), 3.10 (m, 1H), 3.19 (s, 3H), 3.42–3.65 (m, 3H), 3.68 (s, 3H), 3.83 (m, 1H), 4.51 (m, 1H), 7.32 (m, 3H), 7.44 (t, 1H), 8.25 (d, 1H); ¹⁹F NMR δ (84.5 MHz; CD₃OD) -74.3 (s, 3F). 7·HCl: ¹H NMR & (CD₃OD) 2.32 (s, 3H), 2.83 (t, 1H), 3.14 (m, 1H), 3.18 (s, 3H), 3.41–3.63 (m, 3H), 3.50 (s, 3H), 3.81 (m, 1H), 4.42 (m, 1H), 7.07 (d, 1H), 7.16 (d, 1H), 7.24 (d, 1H), 7.40 (t, 1H), 8.36 (d, 1H). 9: ¹H NMR δ (CDCl₃) 1.85 (m, 1H), 2.06 (m, 1H), 2.18 (s, 3H), 2.25-2.34 (m, 2H), 2.44 (s, 3H), 2.59 (m, 1H), 2.67 (m, 1H), 3.06 (d, 1H), 3.34 (m, 1H), 4.17 (m, 1H), 4.84 (d, 1H), 5.29 (m, 1H), 5.68 (m, 1H), 6.54 (d, 1H), 6.82 (d, 1H); IR v/cm⁻¹ (film) 3540 (OH).

‡ All palladium complexes used herein are commercially available.

amide (DMF) according to Stille *et al.*⁷ gave 10-methyl-11methoxyaporphine **7** {hydrochloride, m.p. 189–191 °C, $[\alpha]_D^{21}$ -179.0 (*c* 1.0, MeOH)}. The deoxygenated **8**²⁻⁴ was obtained from **6** by treatment with formic acid (HCO₂H), triethylamine (Et₃N), palladium(11) acetate [Pd(OAc)₂] and 1,1'-bis(diphenylphosphino)ferrocene (dppf) in DMF.⁵ Methoxy derivatives **7** and **8** were *O*-demethylated by treatment with aqueous







 $\begin{array}{l} \textbf{Scheme 2 Reagents and conditions: i, Me_4Sn (1.2 equiv.), LiCl (3.0 equiv.), Pd(PPh_3)_4 (0.050 equiv.), DMF, 100 °C, scaled tube, 18 h, 88%; ii, HCO_2H (3.0 equiv.), Bu_3N (4.2 equiv.), dppp (0.15 equiv.), (PPh_3)_2PdCl_2 (0.060 equiv.), DMF, N_2, 80 °C, 18 h, 91%; iii, MeSO_3H, N_2, 95 °C, 1 h, 82%; iv, MeSO_3H, N_2, 95 °C, 15 min, 67\% \end{array}$

48% hydrobromic acid (HBr) to give 2 and 3¶ in overall yields of 36 and 34%, respectively.

In the second approach (Scheme 2) **5** was converted to **9** {m.p. 181–183 °C, $[\alpha]_D^{21}$ –195.0 (*c* 1.0, MeOH)} by a palladium-catalysed coupling with tetramethyltin as described above for compound **7**. Monotrifluoromethanesulfonate **5** was also used as a precursor in the preparation of 3-deoxymorphine **10** || by palladium-catalysed hydrogenolysis in accordance with the recently published procedure by Saá *et al.*;⁸ tributylammonium formate (hydride donor), bis(triphenylphosphine)palladium(II) chloride [PdCl₂(PPh₃)₂]–1,3-bis(diphenylphosphino)propane [(PPh₂)(CH₂)₃(PPh₂)] (catalytic system) and DMF. Acid catalysed rearrangements^{2,4,6} of **9** and **10** afforded **2** and **3**, respectively. The overall yields in the second route were 65 and 55% for **2** and **3**, respectively.

§ Optical rotation $[\alpha]_D^{21} - 103.4 (c \ 1.0, MeOH), \{\text{lit.}^2 [\alpha]_{578}^{26 \cdot 7} - 85.2 (c \ 0.55, MeOH)\}.$

¶ Compound **3** has been reported as the hydrobromide.^{3,4} Data for **3**·HCl: m.p. 270 °C; $[\alpha]_D^{22}$ -71.6 (*c* 1.0, MeOH).

 \parallel 3-Deoxymorphine **10**, obtained by this route, has been synthesized in eight steps and 16% overall yield from **4**.⁹

It is apparent that the rearrangement of the morphine moiety to the aporphine structure occurs readily also in the absence of the 3-positioned morphine oxygen.** This conclusion was further corroborated by the facile acid-catalysed conversion of 3-deoxy-3-phenylmorphine†† to **11** {hydrochloride, m.p. 246–250 °C, $[\alpha]_D^{23}$ –52.0 (*c* 1.0 MeOH)} in 67% yield. Consequently, the methods presented here appear to be useful for production of an array of pharmacologically interesting 10-substituted derivatives of **2**.

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** We are not aware of any report on morphine to aporphine rearrangements of 3-deoxymorphine derivatives.

^{††} Prepared from **5** by cross coupling with phenylboronic acid in presence of Pd(PPh₃)₄, LiCl and 2 mol dm⁻³ aqueous Na₂CO₃ in refluxing dimethoxyethane (DME)–ethanol mixture under nitrogen atmosphere¹⁰ in 80% isolated yield. M.p. 169–171 °C; $[\alpha]_D^{21}$ –49.0 (*c* 1.0, MeOH).